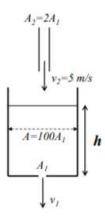
# EXAM PHYSICAL TRANSPORT PHENOMENA – RUG April, 2 - 2019

## Maximum # points for each question are indicated

#### Question 1

Imagine a cylindrical open tank with cross-sectional area A filled with pure water. Because of a hole in the bottom with cross-sectional area  $A_I$  water is flowing out with velocity  $v_I$ . Simultaneously, pure water is flowing in with velocity  $v_2$  through a pipe above the tank with cross-sectional area  $A_2$ , with  $A_2=2A_I$ . Friction losses can be ignored and  $A=100A_I$ , implying  $A>>A_1$ . Gravitational acceleration (g) is  $10 \text{ m/s}^2$ .



A) Calculate the steady-state water level height h in the tank, given  $v_2=5$  m/s.

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While at steady-state (dh/dt=0), at t=0 compound c is added to the inflow stream with concentration  $c_{in}$ . The tank is well-stirred.

B) Give the concentration of c in the tank as function of time given  $c_{in}=10$  mol/l. Check your answer by studying the initial and final concentration of c in the tank!

#### Question 2

Objects moving through fluid experience a force from this fluid, the so-called drag force, measured in Newton with  $1 \text{ N} = 1 \text{ kg.m/s}^2$ . A researcher interested in the swimming behavior and energy consumption of whales aims to perform a study using a scale model, also operating in water. As a start he assumes the drag force to be a function of the velocity of the object, the density of the fluid, the viscosity of the fluid and the diameter of the object. Density and viscosity of water is  $10^3 \text{ kg/m}^3$  and  $10^{-3} \text{ kg/ms}$ , respectively.

A) Apply dimensional analysis and find the two dimensionless groups.

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B) Given the velocity of the prototype during the model study is 10x the (typical) velocity of the whale in real life, what should be the scale factor used?

### Question 3

Consider a rectangular aquarium with a volume of 3 m<sup>3</sup> containing cold-water fish that require the water temperature to be 10  $^{0}$ C. With a (constant) room temperature ( $T_R$ ) of 22  $^{0}$ C, the system needs a cooling element. The water is well stirred. Consider steady-state conditions.

Water density ( $\rho$ ) and specific heat capacity ( $c_p$ ) are  $10^3$  kg.m<sup>-3</sup> and  $4.2.10^3$  J.kg<sup>-1</sup>.K<sup>-1</sup>, respectively.

- A) Given a total average heat transfer coefficient (U) of 420 W/m²K and a total surface area for heat exchange of 6 m², calculate the power of the cooling element.
- B) The aquarium is replaced with a larger one with length, width and height, all twice as large. Given the same steady-state conditions but a heat transfer coefficient of just 300 W/m<sup>2</sup>K (because of the thicker glass!), what should now be the power of the cooling element?
- C) Suppose (for the aquarium in A), the cooling element is turned off. How much time does it take for the water to warm up to  $16^{\circ}$ C?

#### Question 4

One of the threats of accidents on sea is the free-coming oil that can have disastrous effects on the eco-system. Consider a layer of oil floating on the surface of the ocean. Due to diffusion, the oil penetrates into the water, over a period of 1 week to a depth of 75 cm. The concentration of oil at the oil-seawater interface  $(c_s)$  equals the (limit) solubility concentration of oil in seawater,  $10 \text{ g/m}^3$ . For the sake of simplicity the ocean can be considered of endless depth.

- A) Calculate the (apparent) diffusion coefficient (D<sub>A</sub>) of oil in seawater.
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- B) Calculate the average oil concentration in the 75 cm seawater layer after 1 week.

Suppose a similar accident with some chemical happens in a 3 m deep lake resulting in a layer of chemicals floating on the lake surface. Over a period of 1 week, this chemical penetrated to a depth of 2 m, resulting in an (average) concentration in the water of 11 g/m<sup>3</sup>. The concentration of this chemical at the chemical-lake water interface (c<sub>s</sub>) equals the (limit) solubility concentration of this chemical water, 20 g/m<sup>3</sup>.

C) Calculate the (apparent) diffusion coefficient ( $D_A$ ) of this chemical in water.